A novel energetic composite with a special sandwich microstructure: **RDX/expanded graphite intercalation composite**

Xian-Long Ma¹, Shusen Chen¹, Kun Chen¹, Lijie Li¹, Jing-Jing Zhao², Xue-Bin Yang², Fengqin Shang², Xijuan Lv^{*1} and Qinghai Shu^{*1} ¹ School of Materials Science & Engineering, Beijing Institute of Technology. Beijing 100081, PR China ²Research Institute of Gansu Yinguang Chemical Industry Group, Baiyin, PR China. * Corresponding author telephone: Tel: +86-10-68918535; Email: 7520180040@bit.edu.cn; qhshu121@bit.edu.cn

Introduction

- A variety of carbon materials have been developed as carriers of EMs for higher performance.
- Considering that expanded graphite (EG) possesses such many outstanding properties, we employed the solvent/anti-solvent process to recrystallize RDX crystal into holes of EG.
- The advanced sandwich microstructure possessed two effects: (i) ullethigh heat conduction and (ii) hot spots isolation of the carbon microstructure, which were the key to improve thermal stability and decreasing sensitivity.





Preparation of RDX/EG intercalation composite

1) Preparing RDX solution with different concentrations; 2) Adding EG into the solution and soaking to fully wetted; 3) Filtering out the EG absorbing the RDX solution fully from the suspension and pouring the filter residue into deionized water rapidly following the agitation for 5 min;

4) Filtering out the production and washing the sample with deionized water to remove the remaining RDX outside EG;

5) Drying samples at 60°C to the constant weight.



Figure 1. Procedures for the formation of RDX/EG intercalation composites

Figure 5. (a) IR spectroscopies; (b) The XPS general spectroscopy and the C1s, O1s, P2p spectroscopies (in the inserts) of EG; (c) Normalized Raman spectroscopies; (d) XRD patterns

Explosive properties

Sandwich microstructure

Effect of volume ratio of solvent to anti-solvent on intercalation



Figure 2. SEM images of RDX/EG intercalation composites prepared under different volume ratio of solvent to anti-solvent conditions: (a, b) 1:2; (c, d) 1:6; (e, f) 1:10; (g, h) 1:20

Effect of concentration of RDX solution on intercalation





Figure 6. DSC curves of (a) the RDX/EG intercalation composite and (b) raw RDX at different heating rates. Kissinger plots of $\ln(\beta/T_{\rm P}^2)$ versus $1/T_{\rm P}$ were shown in the inserts

Table 1. Kinetics, thermodynamics and thermal stabilities parameters of raw RDX and the maximum intercalated **RDX/EG composite (run 9) derived from their DSC curves**

	P	Ta	Kinetics			Thermodynamics			Thermal stabilities
Samples	β (K·min ⁻¹)	<i>T</i> ₽ (℃)	E_{K}	lgA _K	k	ΔH^{\neq}	ΔG^{\neq}	ΔS^{\neq}	Tb
			(kJ·mol ⁻¹)	(s ⁻¹)	(s ⁻¹)	(kJ·mol ⁻¹)	(kJ·mol ⁻¹)	(J·mol ⁻¹ ·K ⁻¹)	(°C)
	1	222.86			0.10				
Run 9	2	229.46	201.43	20.21	0.19	197.21	131.61	129.24	224.09
	5	239.14			0.47				
	10	246.22			0.89				
	1	215.76			0.08				



Figure 3. SEM images of RDX/EG intercalation composites prepared under different RDX solution concentration conditions: (a, b) 0.10 mol·L⁻¹; (c, d) 0.90 mol·L⁻¹; (e, f) saturation.

Raw RDX	2	223.84	168.06	16.88	0.16	163.88	130.94	65.56	216.44
	5	235.01	108.00		0.40	105.00			210.44
	10	243.03			0.74				

Table 2. Impact and friction sensitivities of raw RDX and **RDX/EG intercalation composites**

Run	Concentration of RDX solution (mol·L ⁻¹)	Volume ratio of solvent to anti- solvent	Weight fraction of RDX in composites (%, Sample-Weight Method)	Impact sensitivities H50 (cm)	Friction sensitivities P (%)
Raw RD2	-	-	-	10.80	72
5	0.9	1:6	74.92	28.40	20
9	1.59 (saturation)	1:6	84.26	27.10	24

Conclusions

The advanced half-coated parallel multi-sandwiches microstructure possessed two effects: (i) heat conduction and (ii) hot spots isolation, which were the key to higher performance.

